

# Ncert Solutions Of Hornbill Class 11th

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## Study Rankers NCERT Solutions for Class 11th Ch 1 Some

January 11th, 2019 - 1 6 Calculate the concentration of nitric acid in moles per litre in a sample which has a density 1 41 g mL<sup>-1</sup> and the mass per cent of nitric acid in it being 69 Answer Mass percent of 69 means that 100g of nitric acid solution contain 69 g of nitric acid by mass

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